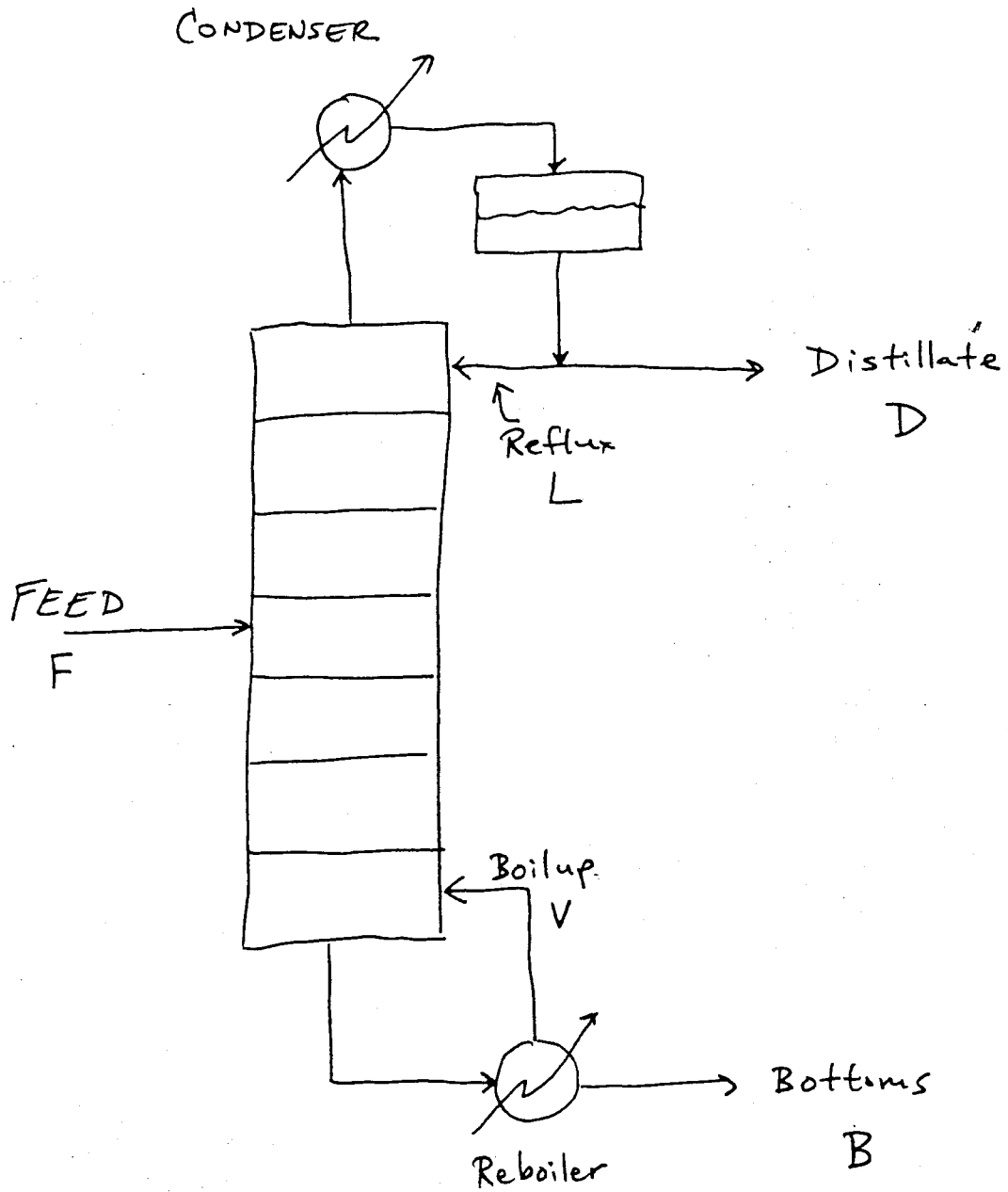


CHAPTER 7: DISTILLATION OF BINARY MIXTURES



WAYS TO RELATE FLOW RATES: (RATIOS)

REFLUX RATIO = $\frac{L}{D}$ $\left\{ \begin{array}{l} \leftarrow \text{Liquid} \\ \leftarrow \text{Distillate} \end{array} \right.$

BOILUP RATIO = $\frac{V}{B}$ $\left\{ \begin{array}{l} \leftarrow \text{Vapor} \\ \leftarrow \text{Bottoms} \end{array} \right.$

FLOW RATES

ANY ASSUMPTIONS HERE?

→ STEADY STATE

→ CONSTANT FLOWS

DO THE ASSUMPTIONS MAKE SENSE?

→ BESIDES WHAT HAPPENS ON THE STAGES...

BIGGEST EVENT TO ALTER L+V?

FEED

FEED CONDITION -

RECALL SINGLE STAGE FLASH

$$\psi = \frac{V}{F} \quad + \quad 0 \leq \psi \leq 1$$

WHY?

⇒ NEEDED 2 PHASES FOR SEPARATION.

NOW, ALREADY HAVE 2 PHASES IN COLUMN.

⇒ FEED CONDITION BECOMES A FACTOR.

POSSIBILITIES :

①

②

③

④

⑤

REFLUX RATIO	BOILUP RATIO

PURE MATERIAL OUT THE TOP \Rightarrow

FOCUS ON REFLUX RATIO

BALANCING ACT

of stages

reflux ratio

① Minimize reflux ratio

→ Liquid only from condensation in each stage.

→ for high purity, # of stages large.

of stages → ∞

② Minimize # of stages

→ NEED LOTS OF LIQUID AT EACH STAGE FOR GOOD MASS TRANSFER

$$\text{Reflux Ratio} = \frac{L}{D}$$

$$L \uparrow \Rightarrow D \downarrow$$

⇒ Reflux Ratio → ∞

MORE ON DISTILLATION...

WOULD LIKE A SYSTEMATIC APPROACH

⇒ MCCABE - THIELE GRAPHICAL TECHNIQUE

① Equilibrium (EQ) Line

② Operating (OP) Line

③ Binary feed mixture

④ Defined column pressure.

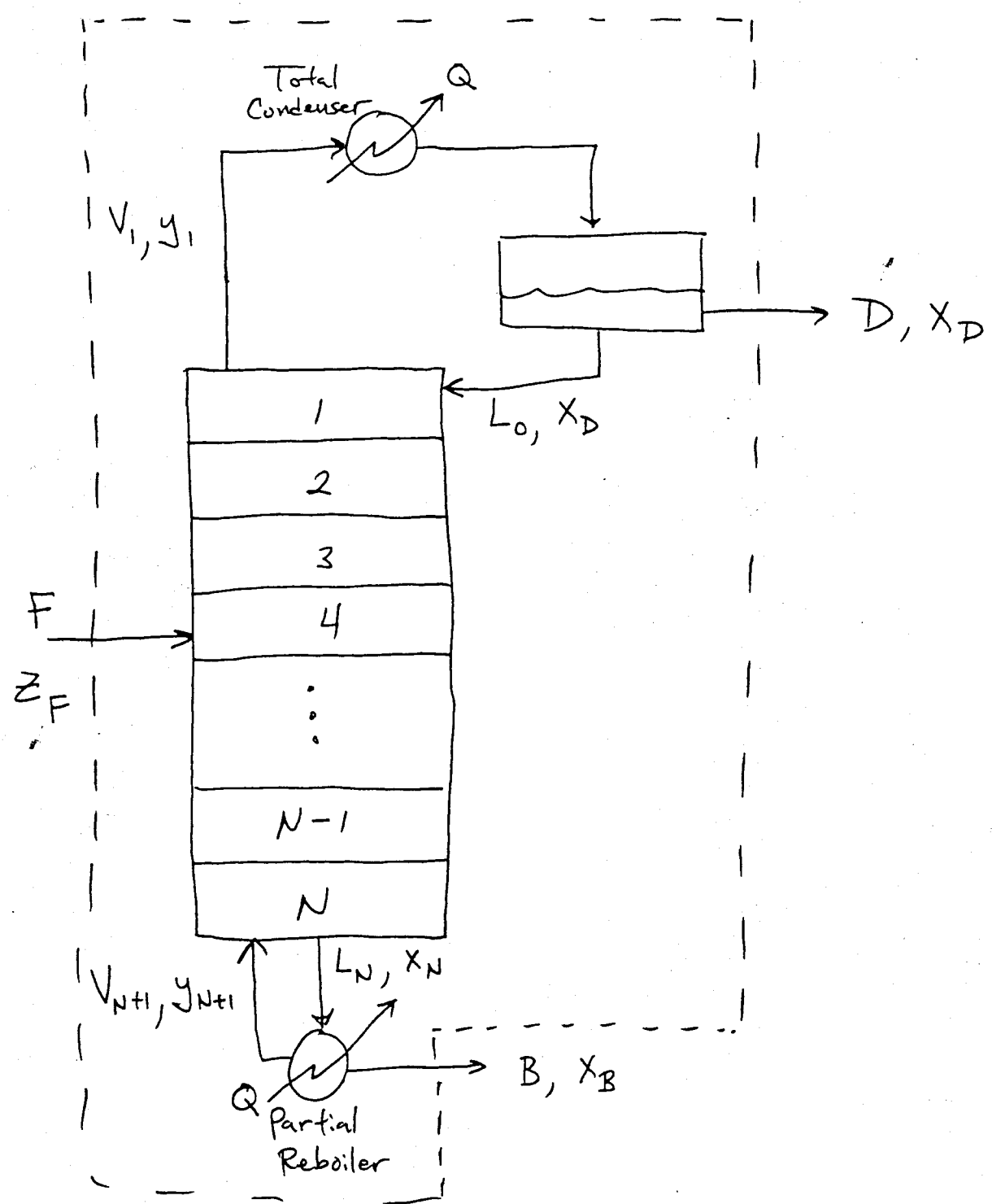
⑤ Allows estimates for:

a) # of EQ Stages

b) amount of reflux for desired separation

⇒ WHAT ARE OPERATING LINES?

BEFORE WE LOOK AT INDIVIDUAL OPERATING LINES, LET'S LOOK AT THE ENTIRE PROCESS



NOTE: MOLE FRACTIONS FOR LIGHT KEY.

⇒ OVERALL MOLE BALANCE

$$\underbrace{F}_{\text{IN}} = \underbrace{D + B}_{\text{OUT}} \quad (1)$$

⇒ COMPONENT MOLE BALANCE (LIGHT KEY)

$$\underbrace{F z_F}_{\text{IN}} = \underbrace{D x_D + B x_B}_{\text{OUT}} \quad (2)$$

COMBINE EQNS (1) + (2) AND SOLVE
FOR D...

$$F z_F = D x_D + (F - D) x_B$$

$$F z_F = D x_D + F x_B - D x_B$$

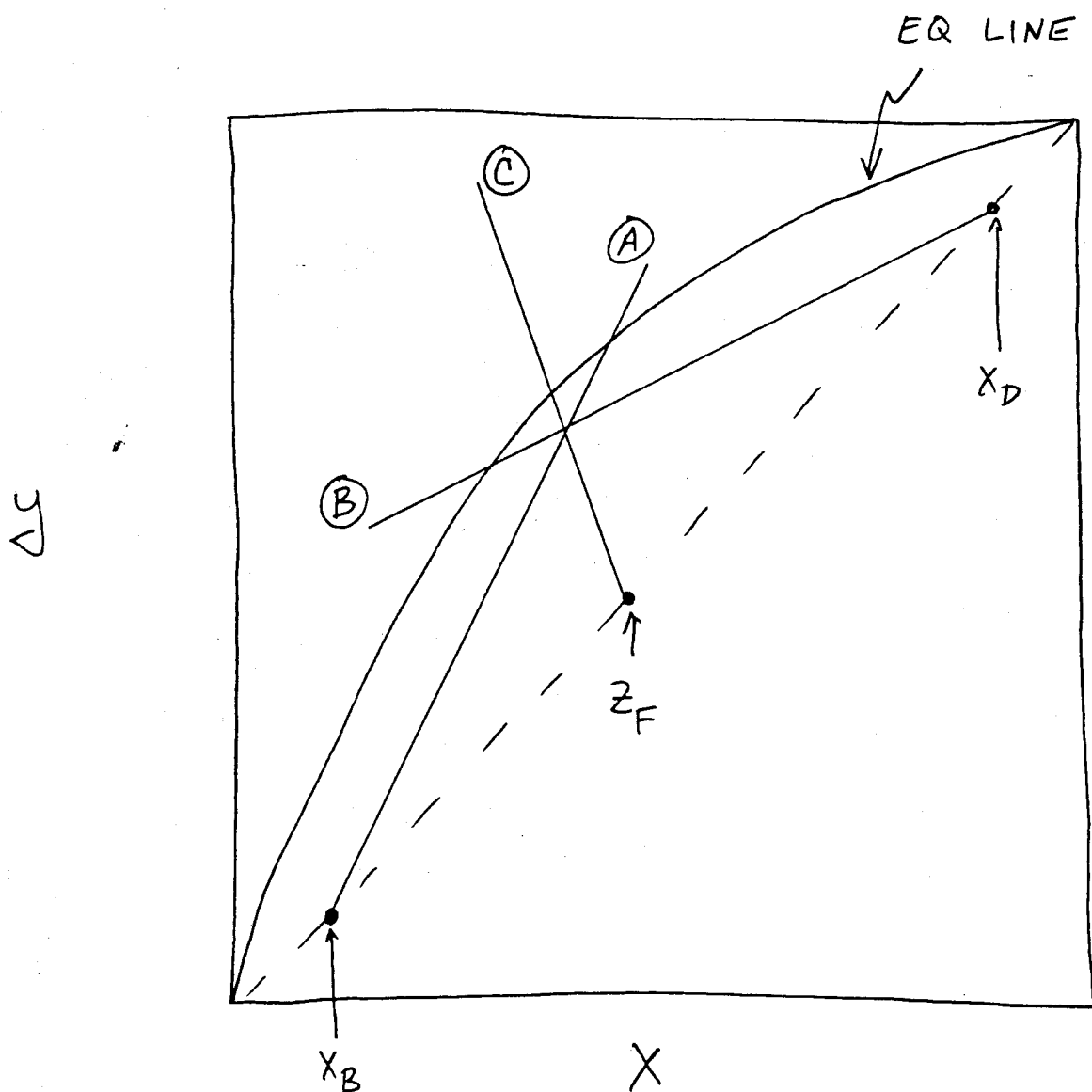
$$D = \frac{F z_F - F x_B}{x_D - x_B} = F \left(\frac{z_F - x_B}{x_D - x_B} \right)$$

WHAT DOES THIS MEAN?

$$X_D > Z_F > X_B$$

⇒ MAKES SENSE. REMEMBER HOW COMPOSITION OF LIGHT KEY CHANGES IN COLUMN.

WELL, WHERE ARE WE HEADED WITH THIS?



(A) STRIPPING SECTION OPERATING LINE

$$\text{Slope} = \frac{\bar{L}}{\bar{V}}$$

(B) RECTIFYING SECTION OPERATING LINE

$$\text{Slope} = \frac{L}{V}$$

(C) q-line

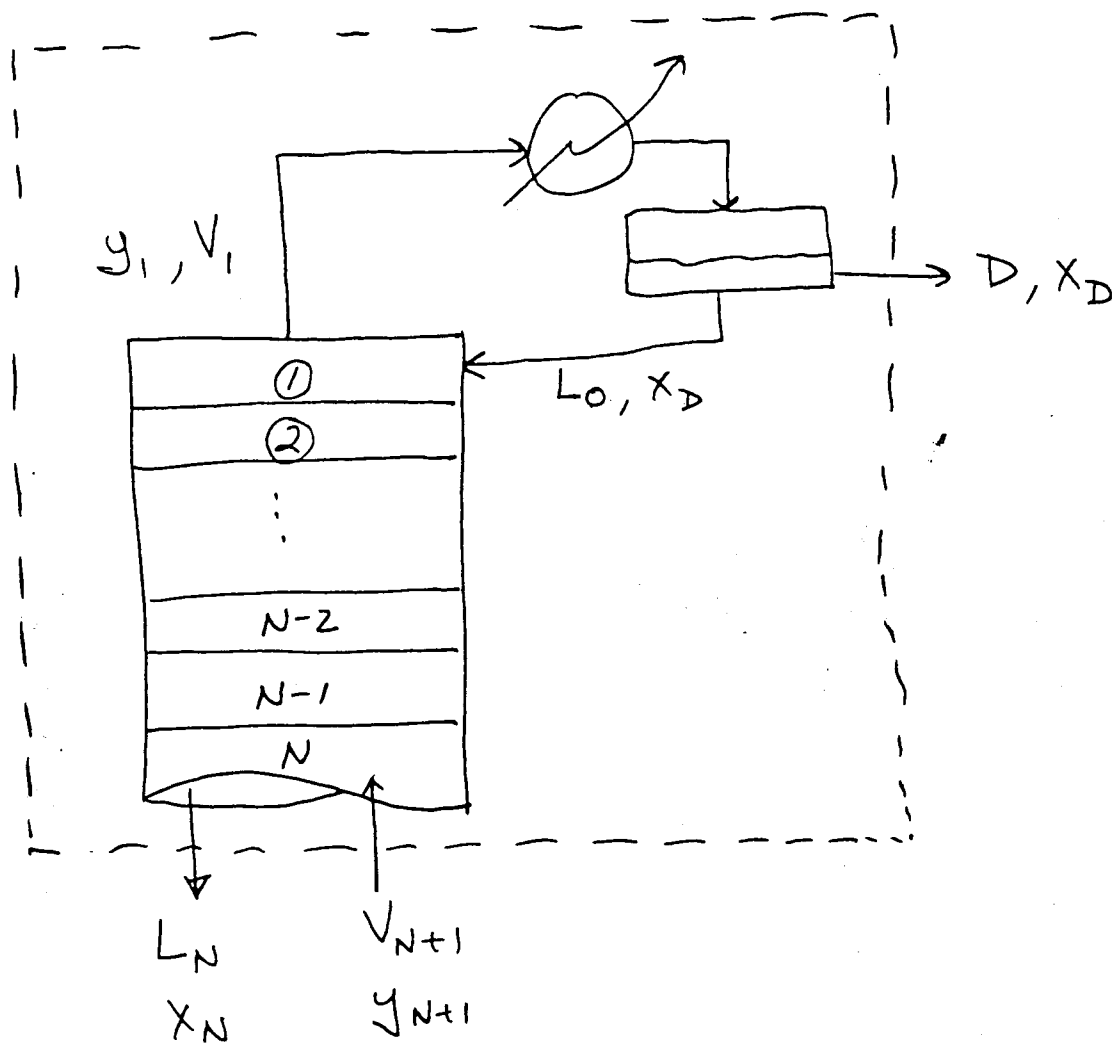
$$\text{Slope} = \frac{\psi - 1}{\psi}$$

Gives location of feed tray

⇒ WHERE DO THESE COME FROM?

⇒ OPERATING LINES...

RECTIFIER OPERATING LINE



COMPONENT MOLE BALANCE (LIGHT KEY)

$$\underbrace{(V_{N+1})(y_{N+1})}_{\text{IN}} = \underbrace{(L_N)(x_N) + (D)(x_D)}_{\text{OUT}}$$

SOLVE FOR y_{N+1} ...

$$y_{N+1} = \left(\frac{L_N}{V_{N+1}} \right) x_N + \left(\frac{D}{V_{N+1}} \right) x_D$$

⇒ LOOKS LIKE DIFFERENT OPERATING LINE FOR EACH STAGE ⇒ SIMPLIFYING ASSUMPTIONS,

ASSUMPTIONS

1) L + V CONSTANT IN RECTIFYING SECTION

⇒ a) No pressure drop

b) $(\Delta H^{vap})_A = (\Delta H^{vap})_B$

c) $(C_p \Delta T)_A + (C_p \Delta T)_B \ll \Delta H^{vap}$

d) column well insulated

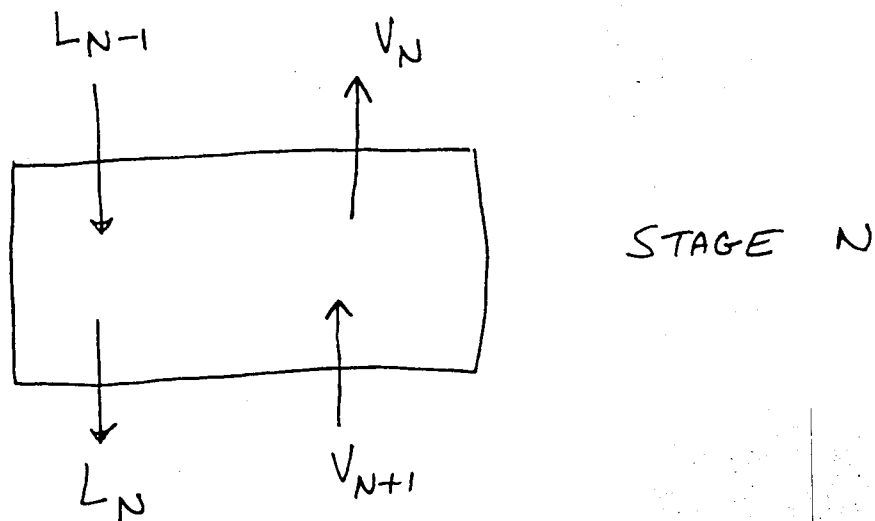
2) Total Condenser ⇒ liquid mole fraction for D.

$$\Rightarrow y_{N+1} = \left(\frac{L}{V}\right) x_N + \left(\frac{D}{V}\right) x_D$$

RECTIFIER OPERATING LINE

\Rightarrow RELATES y_{N+1} and x_N

THESE ARE PASSING STREAMS:



EQUILIBRIUM LINE RELATES COMPOSITIONS
IN $L_N + V_N$

OPERATING LINE RELATES COMPOSITIONS

IN a) L_{N-1} and V_N

b) L_N and V_{N+1}

FROM OPERATING LINE, SLOPE = $\frac{L}{V}$

NEED A POINT TO START AT ...

\Rightarrow 45° Line, $y_{N+1} = x_N$

\Rightarrow Substitute & solve for x_N

$$x_N = \frac{L}{V} x_N + \frac{D}{V} x_D$$

$$x_N \left(1 - \frac{L}{V}\right) = \frac{D}{V} x_D$$

$$x_N = \frac{D}{V} x_D \left(\frac{V}{V-L}\right)$$

But $V = L + D$ (overall mole balance)

$\Rightarrow D = V - L$

$$x_N = \frac{\cancel{D}}{\cancel{V}} x_D \left(\frac{\cancel{V}}{\cancel{D}}\right)$$

$x_N = x_D$

← Starting point on 45° line.

CLOSER LOOK @ SLOPE

FOR RECTIFIER, SLOPE < 1

$$\Rightarrow \underline{L < V \text{ in Rectifier}}$$

$$\frac{L}{V} = \frac{L}{L+D} \quad (\text{overall mole balance})$$

RECALL REFLUX RATIO

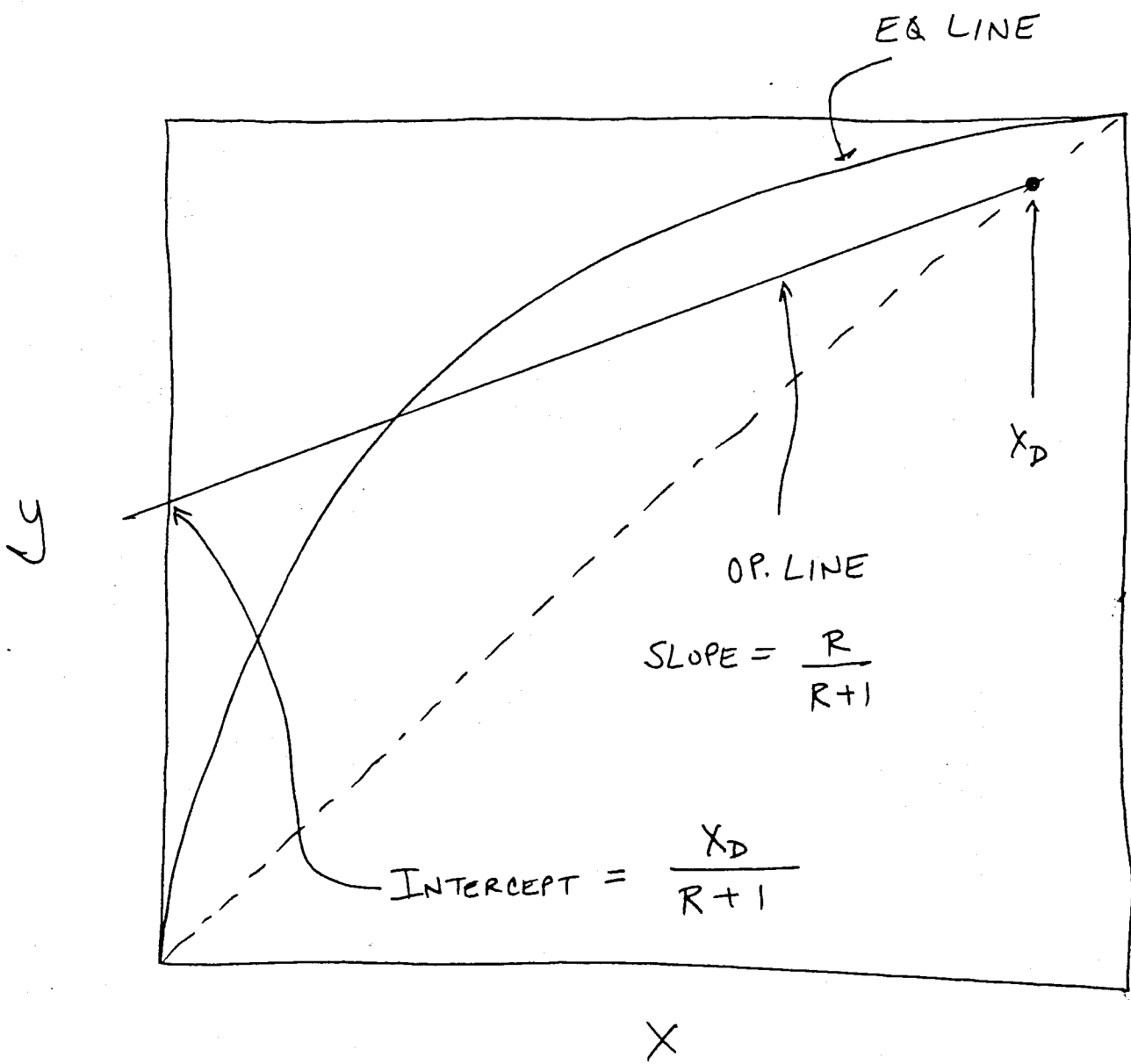
$$\frac{L}{D} = R$$

$$\frac{L}{L+D} = \frac{\frac{1}{D} (L)}{\frac{1}{D} (L+D)} = \frac{\frac{L}{D}}{\frac{L}{D} + 1}$$

$$\Rightarrow \text{Slope} = \frac{L}{V} = \frac{R}{R+1} \quad \text{Rectifier}$$

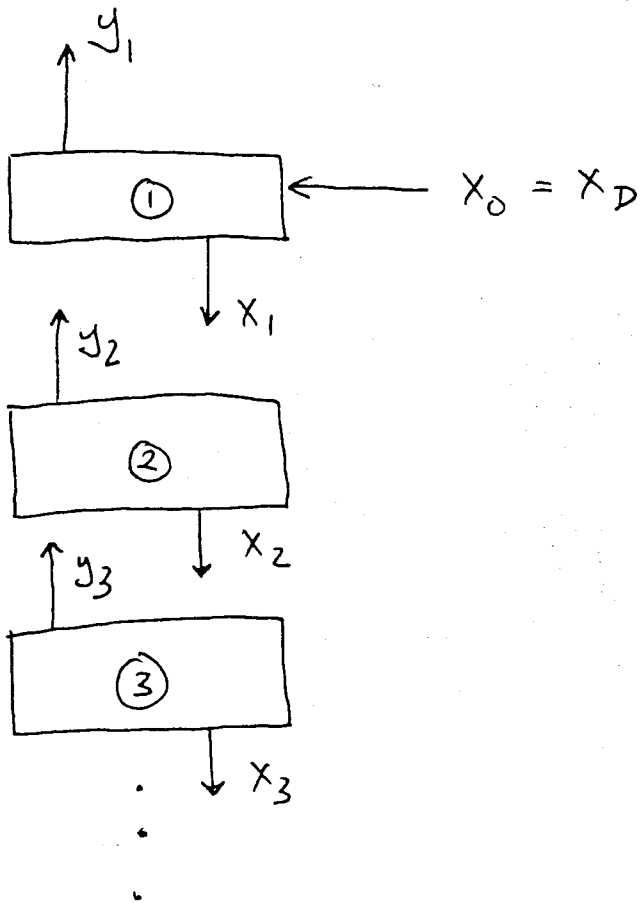
$$\frac{D}{V} = \frac{D}{L+D} = \frac{1}{\frac{L}{D} + 1} = \frac{1}{R+1} \quad \text{INTERCEPT}$$

$$y_{N+1} = \frac{R}{R+1} x_N + \frac{1}{R+1} x_D \quad \text{OPERATING LINE RECTIFIER}$$



⇒ NOW THAT WE HAVE THE OPERATING LINE, HOW DO WE USE THIS FOR DESIGN?

⇒ STEPPING OFF STAGES



PROCEDURE:

- 1) KNOW SPEC. ($X_D = X_0$)
 - 2) Total Condenser $\Rightarrow X_D = y_1$
AND y_1 on ΣQ LINE
 - 3) y_1 in ΣQ w/ x_1 (ΣQ Line)
 - 4) x_1 passing y_2 (OP Line)
 - 5) y_2 in ΣQ w/ x_2 (ΣQ Line)
- etc. (until ΣQ Line + OP Line touch)

EXAMPLE:

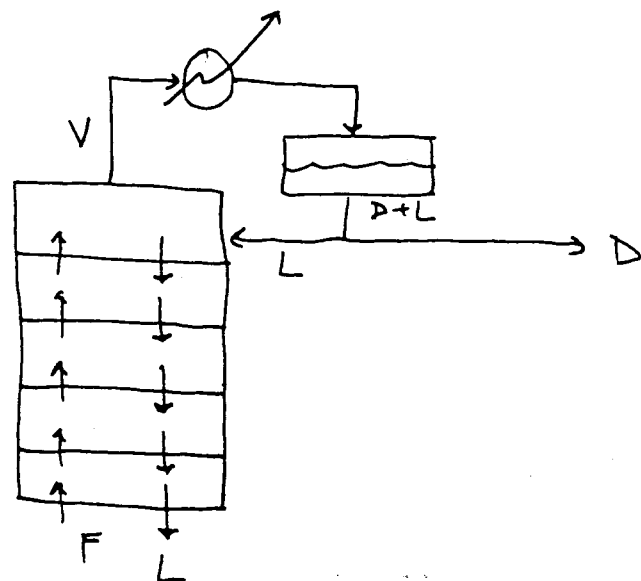
Hexane (1) / Octane (2)

- Spec is 95% hexane
- Total condenser
- Reflux ratio = 2
- Uniform pressure of 1 bar
- XY phase diagram given
- Feed = 100 moles/hour (all vapor)
- Rectifier

⇒ How Good CAN WE DO?

SOLUTION:

Step 1 Picture



STEP 2 WHAT DO WE KNOW?

14

$$\rightarrow X_D = 0.95$$

\rightarrow Total condenser \Rightarrow Not a stage

$$\rightarrow R = \frac{L}{D} = 2$$

$$\rightarrow F = 100 \text{ mol/hr}$$

$$\rightarrow \text{Pressure} = 1 \text{ bar}$$

STEP 3 BULK FLOWS

$$\frac{L}{D} = 2 \quad (\text{Reflux Ratio})$$

$$F = L + D \quad (\text{Overall mole balance})$$

$$F = 100 \quad (\text{given})$$

$$\Rightarrow L = 2D$$

$$100 = 2D + D$$

\Rightarrow

$$D = 33.3 \text{ mol/hr}$$

$$L = 66.7 \text{ mol/hr}$$

STEP 4 BALANCE AROUND DRUM

$$V = D + L = F$$

$$\Rightarrow V = 100 \text{ mol/hr}$$

STEP 5 OPERATING LINE

$$X_D = 0.95 \quad (\text{Starting point on } 45^\circ \text{ line})$$

$$\text{Slope} = \frac{R}{R+1} = \frac{2}{2+1} = \frac{2}{3}$$

(Note: Double check slope < 1)

STEP 6 ONTO XY DIAGRAM.

(see next page)

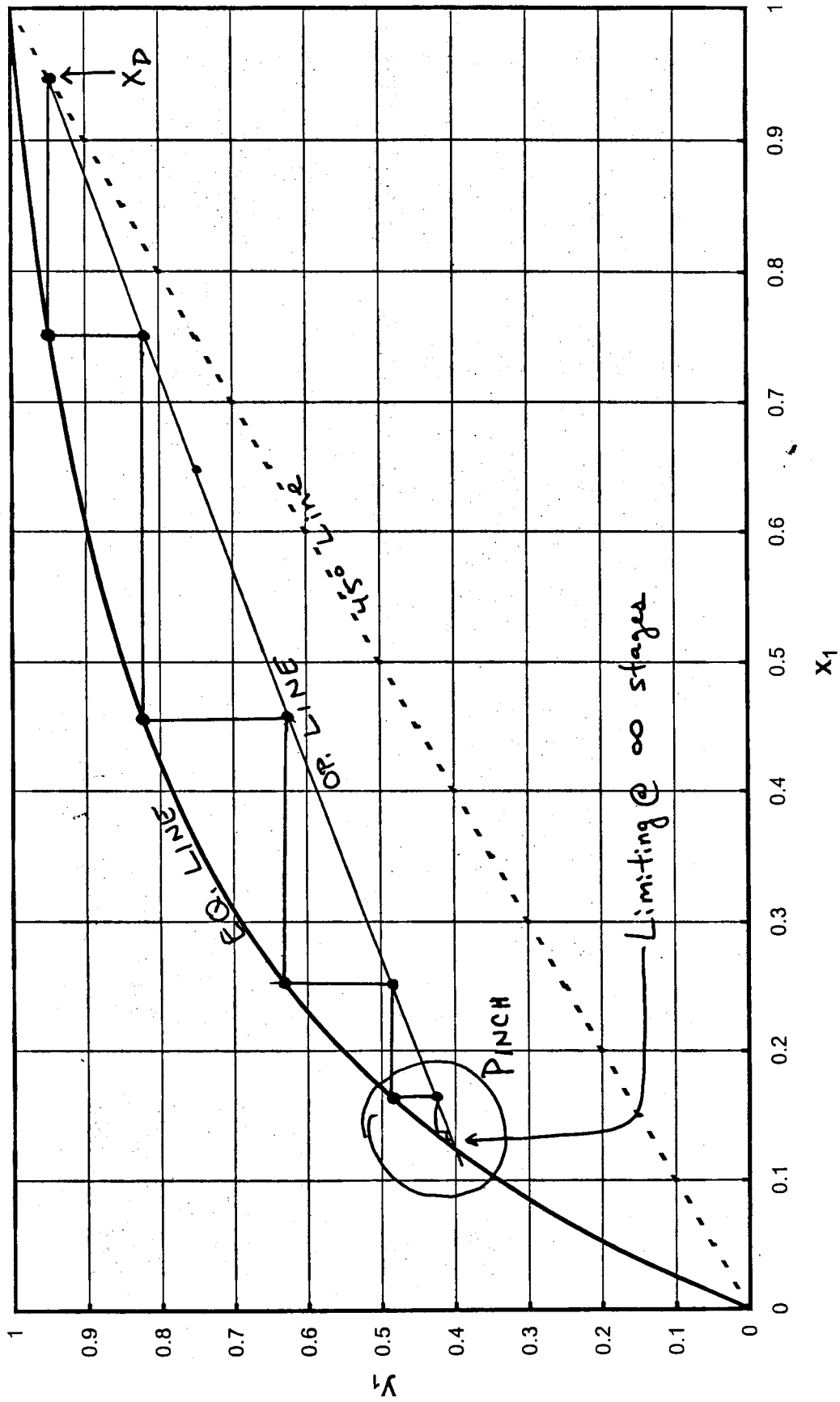
STEP 7 STEP OFF STAGES

- ① $X_D = 0.95$ (spec)
- ② $y_1 = X_D = 0.95$ (total condenser)
- ③ $x_1 = 0.75$ (EQ Line)
- ④ $y_2 + x_1$ passing streams (OP Line)
 $y_2 = 0.825$
- ⑤ $x_2 + y_2$ in EQ (EQ Line)
 $x_2 = 0.45$

and onward ...

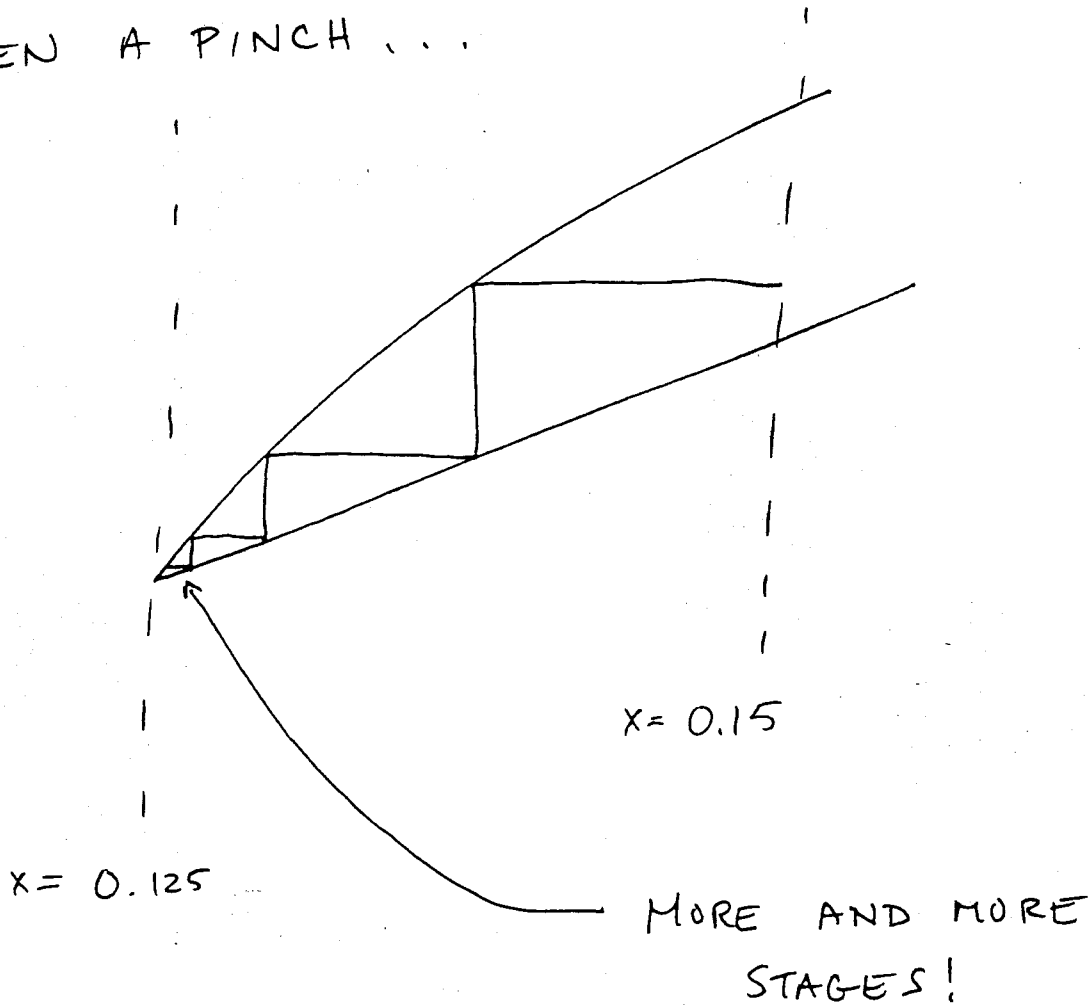
RECTIFIER ONLY

XY Phase Diagram for a System of Hexane (1) in Octane (2) at 1 bar



FOUR STAGES GOT TO $x_4 = 0.16$

THEN A PINCH ...



⇒ DESIGN CHANGE

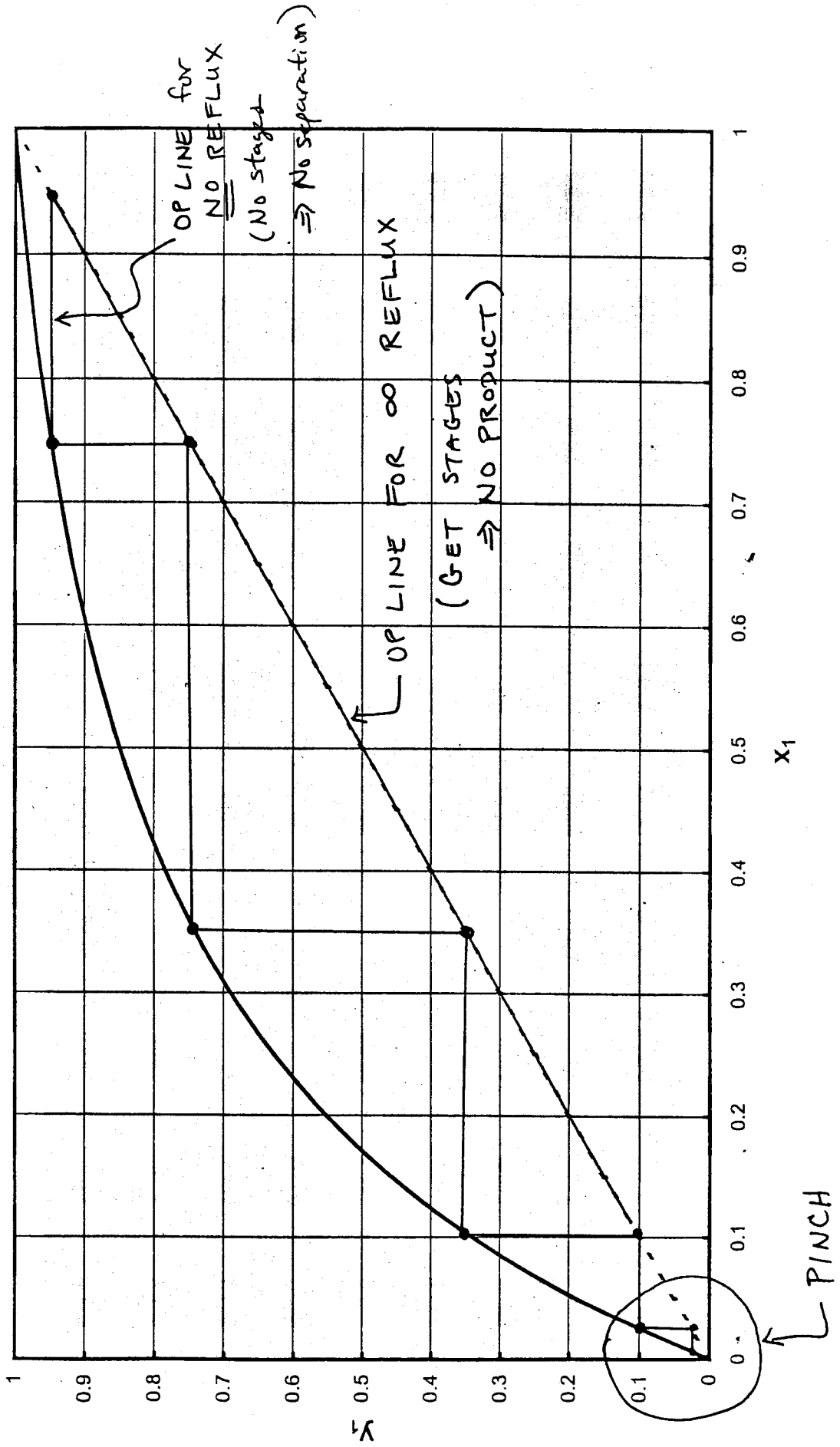
REFLUX RATIO

a) R smaller $\Rightarrow \frac{R}{R+1} \rightarrow 0$

b) R larger $\Rightarrow \frac{R}{R+1} \rightarrow 1$

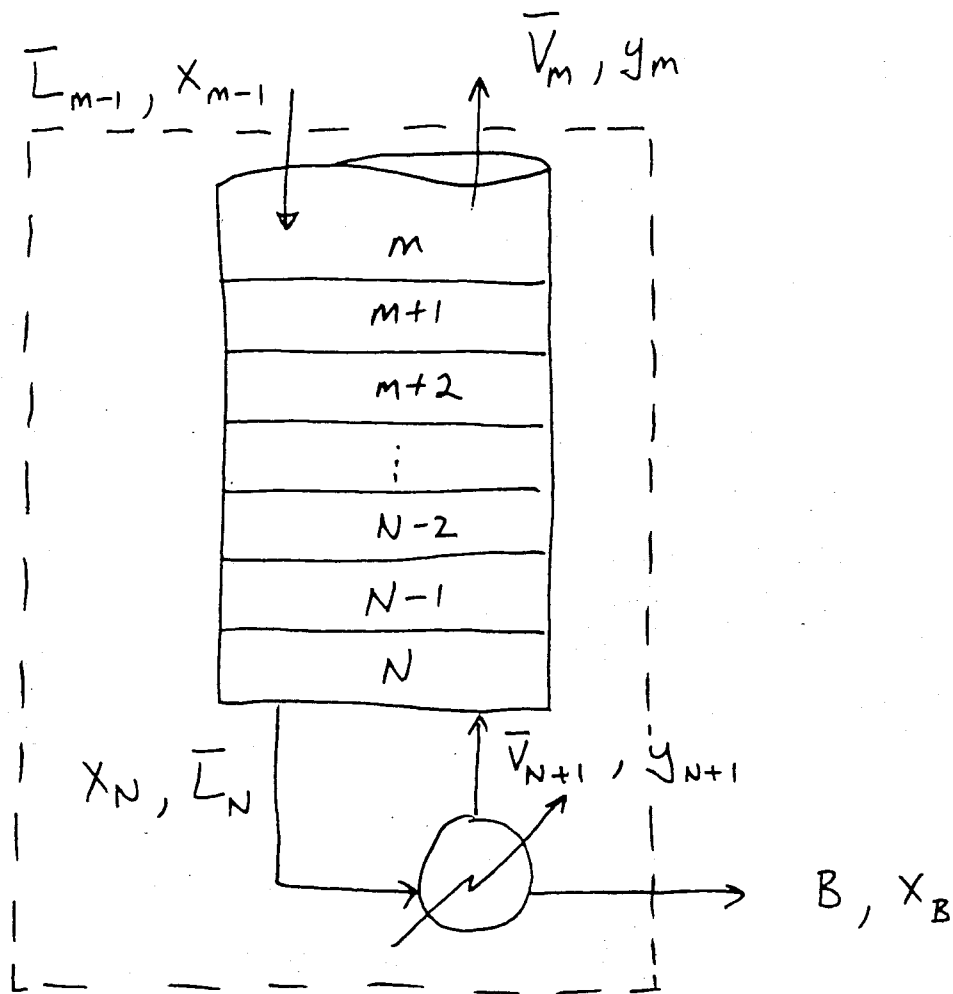
RECTIFIER ONLY

XY Phase Diagram for a System of Hexane (1) in Octane (2) at 1 bar



⇒ NEED BALANCE

STRIPPING SECTION OPERATING LINE



OVERALL MOLE BALANCE!

$$\underbrace{\bar{L}_{m-1}}_{IN} = \underbrace{\bar{V}_m + B}_{OUT}$$

(2)

COMPONENT MOLE BALANCE (LIGHT KEY):

$$\underbrace{(\bar{L}_{m-1})(x_{m-1})}_{IN} = \underbrace{(\bar{V}_m)(y_m) + (B)(x_B)}_{OUT}$$

JUST LIKE FOR RECTIFIER, SOLVE FOR y_m

$$y_m = \frac{\bar{L}_{m-1}}{\bar{V}_m} (x_{m-1}) - \frac{B}{\bar{V}_m} (x_B)$$

⇒ AGAIN, DIFFERENT OPERATING LINE FOR EACH STAGE ⇒ SIMPLIFYING ASSUMPTIONS

ASSUMPTIONS:

1) $\bar{L} + \bar{V}$ constant.

$$\Rightarrow \frac{\bar{L}_{m-1}}{\bar{V}_m} = \frac{\bar{L}}{\bar{V}}$$

2) Bottoms liquid product $\Rightarrow X_B$
(liquid mole fraction)

3) No pressure drop.

$$4) (\Delta H^{vap})_A = (\Delta H^{vap})_B$$

$$5) (C_p \Delta T)_A + (C_p \Delta T)_B \ll \Delta H^{vap}$$

6) Column well insulated.

Slope of
Stripping
Section
Op. Line

$$= \frac{\bar{L}}{\bar{V}}$$

NOTE: SLOPE > 1

$\Rightarrow \bar{L} > \bar{V}$ in STRIPPING SECTION

$$\Rightarrow y_m = \frac{\bar{L}}{\bar{V}} (X_{m-1}) - \frac{B}{\bar{V}} X_B$$

STRIPPER OPERATING LINE

④

CLOSER LOOK @ SLOPE:

$$\frac{\bar{L}}{\bar{V}} = \frac{\bar{V} + B}{\bar{V}} \quad (\text{overall mole balance})$$

RECALL BOILUP RATIO

$$V_B = \frac{\bar{V}}{B}$$

$$\Rightarrow \frac{\frac{1}{B} (\bar{V} + B)}{\frac{1}{B} \bar{V}} = \frac{\frac{\bar{V}}{B} + 1}{\frac{\bar{V}}{B}}$$

$$\Rightarrow \boxed{\frac{\bar{L}}{\bar{V}} = \frac{V_B + 1}{V_B}} \quad \text{SLOPE}$$

$$\boxed{\frac{B}{\bar{V}} = \frac{1}{V_B}} \quad \text{INTERCEPT.}$$

$$\boxed{y_m = \left(\frac{V_B + 1}{V_B} \right) (x_{m-1}) - \frac{1}{V_B} x_B} \quad \text{STRIPPER OP. LINE.}$$